

The Entropy of a
perfect gas
Sakur-Tetrode

(1)

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thermodynamic limit

$$\ln \Gamma(E) \approx \ln \Omega(E)$$

Correct
Boltzmann's
Country

$$\Gamma(E) = \int d^{3N}p d^{3N}q \Delta \delta(E-H) \rightarrow \begin{cases} = 1 & E \leq H \leq E + \Delta \\ = 0 & \text{otherwise} \end{cases}$$

$$\Omega(E) = \int d^{3N}p d^{3N}q \Theta(E-H) \rightarrow \begin{cases} = 1 & H \leq E \\ = 0 & \text{otherwise} \end{cases}$$

Perfect gas

$$H = \sum_i \frac{p_i^2}{2m} \text{ indep on } q \text{ but } \underline{\text{conformant}}$$

$$\Omega(E) = V^N \int d^{3N}p \Theta\left(E - \sum_i \frac{p_i^2}{2m}\right)$$

volume of 3N dimensional sphere
of radius $R^2 = 2mE$

$$\Omega(E) = V^N \Omega_{\substack{3N \\ d=3N}} \int_0^R dp p^{d-1} = V^N \Omega_{3N} \frac{R^d}{d}$$

Calculation of d-dimensional solid angle

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$$\int d^d x e^{-\sum_i x_i^2} = \left(\int dx e^{-x^2} \right)^d = (\sqrt{\pi})^d$$

in spherical coordinates

$$\Omega_d \int_0^\infty dx x^{d-1} e^{-x^2} = \frac{\Omega_d}{2} \int_0^\infty dt t^{\frac{d}{2}-1} e^{-t} = \frac{\Omega_d}{2} \Gamma(d/2)$$

$\xrightarrow{x^2=t}$

The Γ function

$$\Gamma(\alpha) = \int_0^\infty dt t^{\alpha-1} e^{-t} = (\alpha-1)!$$

$$\Omega_d = \frac{2 \pi^{d/2}}{\Gamma(d/2)}$$

in large dimensions the volume of a sphere is

$$\Omega_d \frac{R^d}{d} = \frac{2 \pi^{d/2}}{d \Gamma(d/2)} R^d = \frac{\pi^{d/2}}{(d/2)!} R^d \quad \frac{d}{2} \Gamma\left(\frac{d}{2}\right) = \frac{d}{2}!$$

$$\Omega(E) = V^N \frac{\pi^{3N/2}}{(3N/2)!} (2mE)^{3N/2}$$

large N

$$\left(\frac{3N}{2}\right)! \approx \left(\frac{3N}{2}\right)^{\frac{3N}{2}} e^{-\frac{3N}{2}} \sqrt{2\pi \frac{3N}{2}}$$

Stirling's

$$\Omega(E) = V^N \left(\frac{2\pi m E}{\frac{3N}{2}} e \right)^{\frac{3N}{2}} \frac{1}{\sqrt{2\pi \frac{3N}{2}}}$$

the Boltzmann's entropy

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$$S = k_B \ln \frac{\Omega(E)}{\Lambda^{3N}} = k_B \ln V^N \underbrace{\left(\frac{4\pi m E}{\Lambda^2 3N} e \right)^{\frac{3N}{2}}}_{\text{extensive}}$$

$$+ \frac{k_B}{2} \ln 2\pi \frac{3N}{2}$$

sub extensive

$$S = \underbrace{k_B N \ln V}_{\text{sup extensive}} + \underbrace{\frac{3N}{2} k_B \ln \left[\frac{4\pi m E}{3N \Lambda^2} e \right]}_{\text{extensive}} \quad \text{intensive}$$

Correct Boltzmann's Counting

$$S = k_B \ln \frac{\Omega(E)}{\Lambda^{3N} N!} \quad \text{permutation of identical particles}$$

$$S = k_B \ln \left\{ \frac{V^N}{N!} \left[\frac{4\pi m E}{3N \Lambda^2} \right]^{\frac{3N}{2}} \right\}$$

Stirling

$$\frac{V^N}{N!} = \frac{V^N}{N^N} e^N \left(\frac{1}{\sqrt{2\pi N}} \right)$$

sub extensive into logarithm

intensive

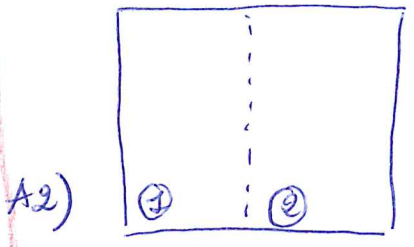
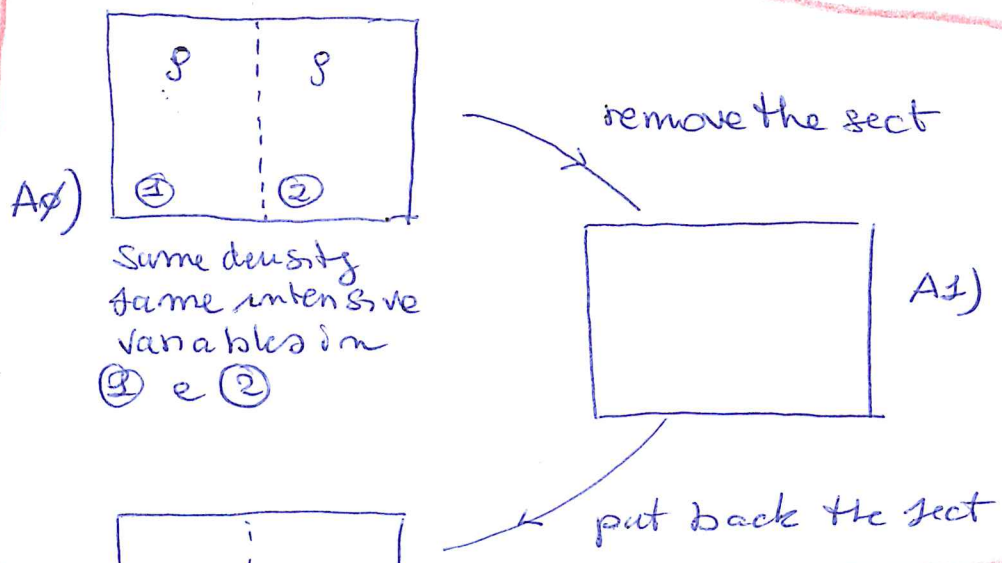
$$S = k_B N \ln \left[\frac{V}{N} e \left(\frac{4\pi m E}{3N h^2} \right)^{3/2} \right]$$

Sakur-Tetrode

Correct Boltzmann's Counting

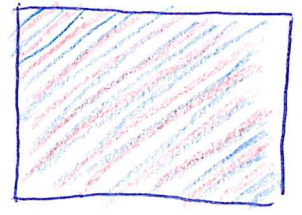
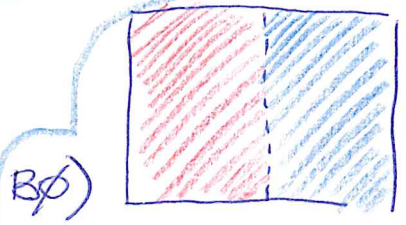
Thermodynamic perspective

Same Gas

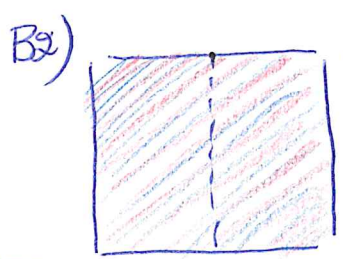


$$A\phi \equiv A2$$

different gases



B1)



$$B\phi \neq B2$$

Consider the entropy without CCB

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$$S' = k_B N \ln V + \frac{3N}{2} k_B \ln \left(\frac{4\pi m E}{3N\lambda^2} e \right)$$

$$S' = k_B N \ln V + \frac{3N}{2} k_B \ln \frac{E}{N} + \underbrace{\text{const} \frac{3N}{2} k_B \ln \frac{4\pi m e}{3N\lambda^2}}_{S'}$$

$$S(A1) = k_B N \ln V + \frac{3}{2} N \ln \frac{E}{N} + S'$$

$$N = N_1 + N_2 \quad E = E_1 + E_2$$

but all the intensive variables are the same

$$S(A0) = k_B N_1 \ln V_1 + k_B N_2 \ln V_2 + \frac{3N_1}{2} k_B \ln \frac{E_1}{N_1} + \frac{3N_2}{2} k_B \ln \frac{E_2}{N_2} + S'$$

$$\frac{E}{N} = \epsilon = \frac{E_1}{N_1} = \frac{E_2}{N_2}$$

$$S(A1) = k_B (N_1 + N_2) \ln (V_1 + V_2) + \frac{3}{2} (N_1 + N_2) \ln \epsilon + S'$$

$$S(A0) = k_B N_1 \ln V_1 + k_B N_2 \ln V_2 + \frac{3}{2} (N_1 + N_2) \ln \epsilon + S'$$

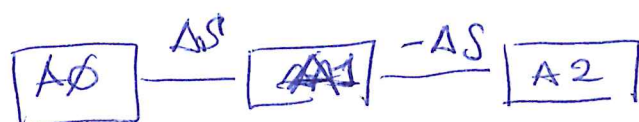
$$\Delta S' = S(A1) - S(A0) = k_B N_1 \ln \frac{V_1 + V_2}{V_1} + k_B N_2 \ln \frac{V_1 + V_2}{V_2}$$

gas 1
diffuses
in 1+2

gas 2 diffuses
in 1+2

if $A\phi \equiv A2$ then

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since $A\phi = A2$ $S(A2) = S(A\phi)$

where is the negative entropy induced by the insertion of a sect from $A1$ to $A2$

?

indeed what is calculated is the MIXING ENTROPY which is correct in the B case from Bφ to B1 for gas which are different (but with particles of equal masses)

Consider the same calculation with CCB

$$S' = k_B N \ln \left(\frac{V}{N} \right) + k_B N \ln \left(\frac{E}{N} \right) + S'$$

intensive

$$S'(A\phi) = k_B (N_1 + N_2) \ln \frac{V_1 + V_2}{N_1 + N_2} + k_B (N_1 + N_2) \ln \frac{E_1 + E_2}{N_1 + N_2} + S'$$

but $\frac{V_1 + V_2}{N_1 + N_2} = \frac{V_1}{N_1} = \frac{V_2}{N_2} = v$ volume per particle

$\frac{E_1 + E_2}{N_1 + N_2} = \frac{E_1}{N_1} = \frac{E_2}{N_2} = \epsilon$ energy x particle

therefore

$$S(A\phi) = S_1 + S_2 = S'(A1)$$

for a system of 2 kind of particles

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$$\mathcal{Z} = k_B \ln \frac{\Gamma_{1+2}(E)}{N_1! N_2! \Lambda^{3N_1+3N_2}}$$

↑
different from $(N_1+N_2)!$

$$\Gamma_{1+2}(E) = \Delta \int d^{3N_1} p_1 d^{3N_1} q_1 \int d^{3N_2} p_2 d^{3N_2} q_2 \delta(E - H_{1+2})$$

for a mixture of 2 perfect gases

$$\mathcal{Z} = \frac{V_1^{N_1} V_2^{N_2}}{N_1! N_2!} \left(\frac{4\pi m_1 E_1}{3N_1 h^2} \right)^{\frac{3N_1}{2}} \left(\frac{4\pi m_2 E_2}{3N_2 h^2} \right)^{\frac{3N_2}{2}}$$

$$S = S_1 + S_2$$