

fugacity expansion for classical gases $z = e^{\beta\mu}$ ①

$$\mathcal{Z} = \sum_{N=0}^{\infty} z^N \mathcal{Z}_N = 1 + z \mathcal{Z}_1 + z^2 \mathcal{Z}_2 + \mathcal{O}(z^3)$$

$$\ln \mathcal{Z} = z \mathcal{Z}_1 + z^2 \mathcal{Z}_2 - \frac{z^2 \mathcal{Z}_1^2}{2} = \cancel{z \mathcal{Z}_1} + \frac{z^2 \mathcal{Z}_2}{2}$$

$$\Omega = -PV = - \frac{(z - \frac{z^2}{2}) \mathcal{Z}_1 + z^2 \mathcal{Z}_2}{\beta}$$

$$\langle N \rangle = - \frac{\partial \Omega}{\partial \mu} = - \frac{\partial \Omega}{\partial z} \frac{\partial z}{\partial \mu} = + \frac{1}{\beta} \left(\frac{\partial \ln \mathcal{Z}}{\partial z} \right) \beta z = z \frac{\partial \ln \mathcal{Z}}{\partial z}$$

$$\langle N \rangle = z (\mathcal{Z}_1 + 2z \mathcal{Z}_2 - z \mathcal{Z}_1^2)$$

$$\frac{\beta P}{P} = \frac{(z - \frac{z^2}{2}) \mathcal{Z}_1 + z^2 \mathcal{Z}_2}{z \mathcal{Z}_1 + 2z^2 \mathcal{Z}_2 - z^2 \mathcal{Z}_1^2} = \frac{\mathcal{Z}_1 + z(\mathcal{Z}_2 - \frac{1}{2} \mathcal{Z}_1^2)}{\mathcal{Z}_1 + 2z(\mathcal{Z}_2 - \frac{1}{2} \mathcal{Z}_1^2)} = \frac{1 + z\varepsilon}{1 + 2z\varepsilon}$$

$$\varepsilon = (\mathcal{Z}_2 - \frac{1}{2} \mathcal{Z}_1^2) / \mathcal{Z}_1$$

$$\frac{\beta P}{P} \approx (1 + z\varepsilon)(1 - 2z\varepsilon) = 1 + z\varepsilon - 2z\varepsilon = 1 - z\varepsilon$$

$$\frac{\beta P}{P} = 1 - z \frac{\mathcal{Z}_2 - \frac{1}{2} \mathcal{Z}_1^2}{\mathcal{Z}_1} = 1 + z \mathcal{Z}_1 \left[\frac{1}{2} - \frac{\mathcal{Z}_2}{\mathcal{Z}_1^2} \right]$$

we can eliminate z from this equation

leading to the equation of state

$$z = \frac{\langle N \rangle}{\mathcal{Z}_1}$$

assuming z calculated to first order
one get

②

$$\ln Z = z Z_1$$

$$\langle N \rangle = z \frac{\partial}{\partial z} \ln Z = z Z_1$$

$$\frac{BP}{p} = 1 + \langle N \rangle \left[\frac{1}{2} - \frac{z_2}{z_1^2} \right]$$

therefore the ^{second} ~~first~~ virial coefficient is

$$\frac{BP}{p} = 1 + p \underbrace{V \left[\frac{1}{2} - \frac{z_2}{z_1^2} \right]}_{B_2}$$

Better use

$$z_2 - \frac{1}{2} z_1^2$$

$$B_2 = V \frac{z_1^2 - 2z_2}{2z_1^2} = -\frac{V}{z_1} \frac{z_2 - \frac{1}{2} z_1^2}{z_1} = -\frac{V}{z_1} \epsilon$$

$$Z_1 = \int \frac{d^3q d^3p}{h^3 1!} e^{-\beta \frac{p^2}{2m}} = \frac{V}{\lambda^3} \text{ no ext fields} \quad (3)$$

$$Z_2 = \frac{1}{2!} \int \frac{d^3q_1}{\lambda^3} \frac{d^3q_2}{\lambda^3} e^{-\beta V(q_1 - q_2)}$$

$$Z_2 - \frac{1}{2} Z_1^2 = \frac{1}{2} \int \frac{d^3q_1 d^3q_2}{\lambda^3 \lambda^3} - \frac{1}{2} \int \frac{d^3q_1}{\lambda^3} \frac{d^3q_2}{\lambda^3}$$

$$= \frac{1}{2} \int \frac{d^3q_1 d^3q_2}{\lambda^6} (e^{-\beta V(q_1 - q_2)} - 1) = \frac{V}{2\lambda^6} \int d^3r (e^{-\beta V(r)} - 1)$$

$$\frac{Z_2 - \frac{1}{2} Z_1^2}{Z_1} = \frac{1}{2\lambda^3} \int d^3r [e^{-\beta V(r)} - 1] = \varepsilon$$

eliminate $z = e^{\beta\mu}$ to get the equation of state to first order

$$\ln Z = z Z_1$$

$$\langle N \rangle = z \frac{\partial}{\partial z} \ln Z = z Z_1$$

$$\Rightarrow z = \frac{\langle N \rangle}{Z_1} \text{ therefore}$$

$$\frac{\beta P}{\rho} = 1 - \frac{\langle N \rangle}{Z_1} \varepsilon = 1 - \frac{\langle N \rangle}{V} \lambda^3 \frac{1}{2\lambda^3} \int d^3r (e^{-\beta V(r)} - 1)$$

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$$\frac{\beta P}{\rho} = 1 - \frac{\rho}{2} \int d\mathbf{r} (e^{-\beta \psi(r)} - 1)$$

at low density

$$B_2 = -\frac{1}{2} \int d\mathbf{r} [e^{-\beta \psi(r)} - 1]$$

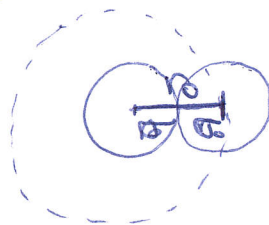
for hard spheres

Mayer function

$$\psi(r) = \begin{cases} = \infty & |r| < r_0 \\ = 0 & \text{otherwise} \end{cases}$$

$$B_2 = +\frac{1}{2} 4\pi \int_0^{r_0} dr r^2 = +\frac{1}{2} \left(\frac{4}{3} \pi r_0^3 \right) \text{ excluded volume volume of the hard sphere}$$

$$\frac{\beta P}{\rho} = 1 + \frac{\rho}{2} \sigma_{\text{excl}}$$



The ^{radius} volume of the sphere is σ then here $r_0 = 2\sigma$

$$B_2 = \frac{1}{2} \frac{4}{3} \pi 8 \sigma^3 = 4 \sigma_{\text{sphere}}$$

The virial coefficients of the hard sphere fluids do not depend on temperature

$$\frac{\beta P}{\rho} = 1 + \rho \underbrace{V \left[- \frac{Z_2 - \frac{1}{2} Z_1^2}{Z_1^2} \right]}_{B_2} + B_3 \rho^2 + B_4 \rho^3 + \dots$$

(5)

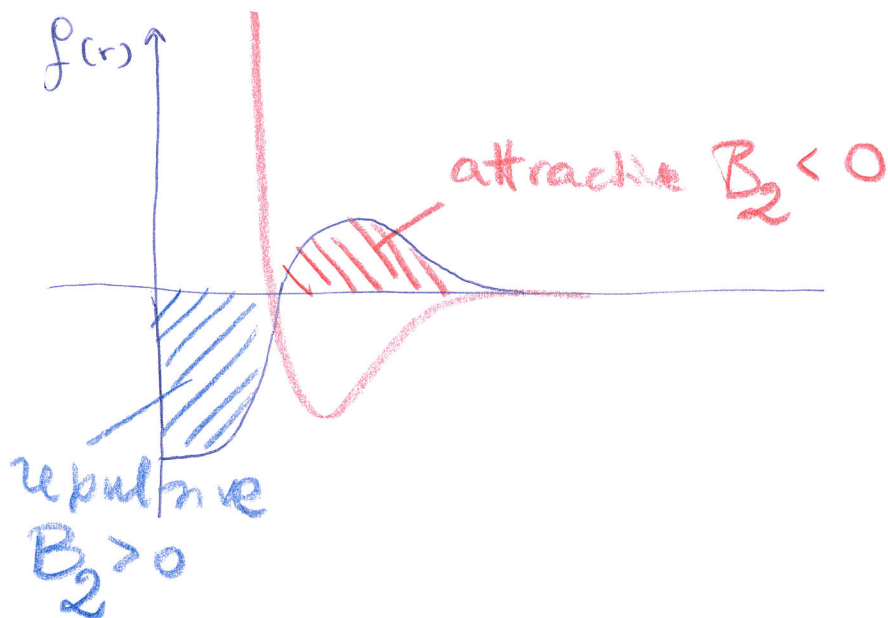
$$Z_1 = \frac{V}{\lambda^3}$$

$$Z_2 = \frac{1}{2!} \int \frac{d^3 q_1 d^3 q_2}{\lambda^6} e^{-\beta \phi} = \frac{V}{2\lambda^6} \int d^3 r e^{-\beta \phi}$$

$$Z_2 - \frac{1}{2} Z_1^2 = \frac{V}{2\lambda^6} \int d^3 r e^{-\beta \phi(r)} - \frac{V^2}{2\lambda^6} = \frac{V}{2\lambda^6} \int d^3 r (e^{-\beta \phi(r)} - 1)$$

$$V \frac{Z_2 - \frac{1}{2} Z_1^2}{Z_1^2} = \frac{\frac{V^2}{2\lambda^6} \int d^3 r (e^{-\beta \phi(r)} - 1)}{\cancel{V^2} \cancel{\lambda^6}} = \frac{1}{2} \int d^3 r$$

$$B_2 = - \frac{1}{2} \int d^3 r \underbrace{(e^{-\beta \phi(r)} - 1)}_{f(r)}$$

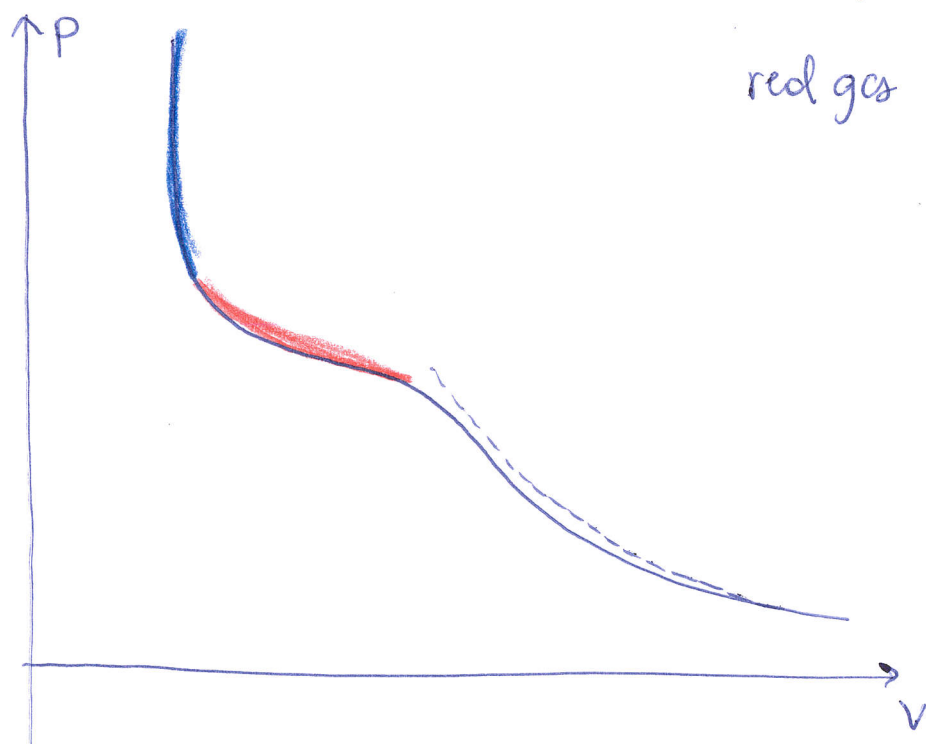
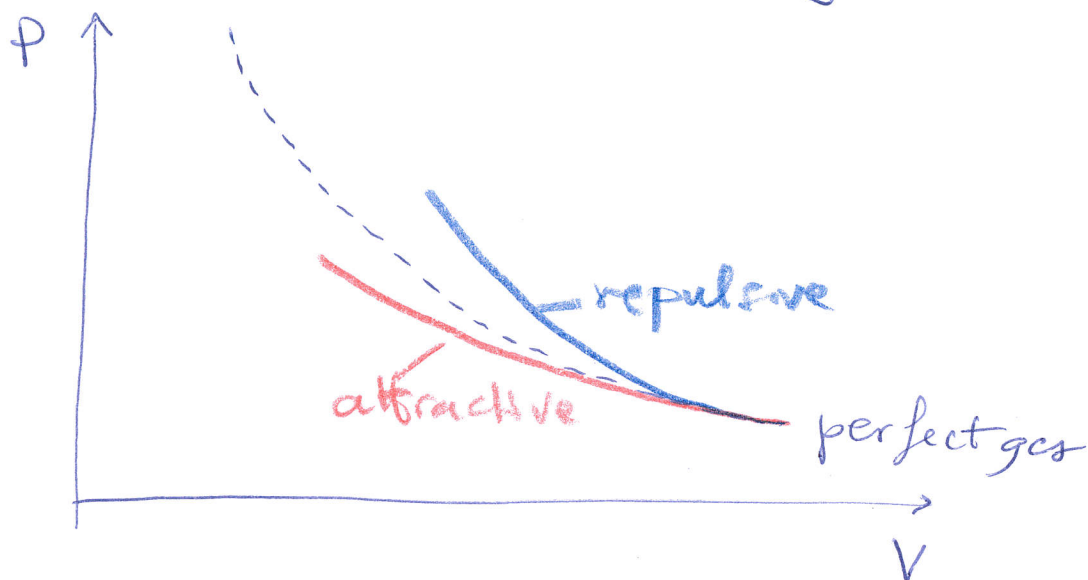


$$\frac{PV}{Nk_B T} = 1 + p B_2$$

$$T = \text{const}$$

$$B_2 = \text{const}$$

⑥



classical phase transition $\rightarrow$ competition
 btw attractive
 repulsive
 interaction